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Key indicators

Single-crystal X-ray study T = 178 K Mean σ (C–C) = 0.003 Å R factor = 0.047 wR factor = 0.138 Data-to-parameter ratio = 13.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-(Dicyanomethylene)-1,5-diphenylpenta-1,4-diyne

The molecule of the title compound, $C_{20}H_{10}N_2$, displays no imposed symmetry, but is largely planar [one phenyl ring is twisted by 10.4 (1)° out of the plane of the rest of the molecule]. The central C=C bond length is 1.371 (3) Å. The molecules are linked by a weak hydrogen bond of the form $C-H\cdots N$.

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Comment

Cross-conjugated enediynes are attracting increasing attention, since they can be used for the preparation of extended π -systems with novel electronic properties (Zhao *et al.*, 2002). We are interested in the synthesis of derivatives of 1,1diethynylethene that bear polarizing functional groups (Hopf *et al.*, 1991). As a reference compound, we prepared the title compound, (I), by the condensation of 1,5-diphenylpenta-1,4diyn-4-one with malonitrile in acetic acid/ethanol in the presence of β -alanine as catalyst (Kreutzer, 1993); we report here its crystal structure.



The molecule (Fig. 1) displays no imposed crystallographic symmetry. It is approximately planar; a closer analysis shows that the ring C9–14 is rotated by 10.4 (1)° from the rest of the molecule (mean deviations 0.004/0.033 Å, respectively, for these two parts). The molecular dimensions may be regarded as normal; the central double bond C3=C6 has a length of 1.371 (3) Å, and the angles subtended by each pair of substituents at this bond are slightly less than the ideal 120° [118.6 (2)° for the phenylethynyl and 117.4 (2)° for the cyano groups]. These values may be compared with the values of 1.378 (3)/1.373(3) Å and 116.8 (1)/118.2(2)° observed in two independent centrosymmetric molecules of tetrakis(phenylethynyl)ethene (Hopf *et al.*, 1991).

The molecules are connected by a weak C19-H19···N1 hydrogen bond by the 2_1 operator parallel to the *b* axis, forming a flattened herring-bone pattern (Fig. 2).

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Figure 1

The molecule of the title compound in the crystal. Ellipsoids represent 50% probability levels.

Experimental

Crystals of (I) were grown by diffusion of pentane into a solution in chloroform.

 $D_x = 1.191 \text{ Mg m}^{-2}$

Cell parameters from 50

 $0.70 \times 0.20 \times 0.15 \text{ mm}$

3 standard reflections

every 147 reflections

intensity decay: none

Mo $K\alpha$ radiation

reflections $\theta = 10-12.5^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$

T = 178 (2) KPrism, yellow

 $\theta_{\rm max} = 25.0^{\circ}$ $h = -13 \rightarrow 14$

 $k = -18 \rightarrow 0$

 $l=-10\rightarrow 0$

Crystal data

$C_{20}H_{10}N_2$
$M_r = 278.30$
Monoclinic, P21/a
a = 12.138 (4) Å
b = 15.279 (4) Å
c = 8.763 (3) Å
$\beta = 107.24 \ (3)^{\circ}$
V = 1552.1 (8) Å ³
Z = 4

Data collection

Nicolet R3 diffractometer ω scans Absorption correction: none 2909 measured reflections 2722 independent reflections 1498 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.050$

Refinement

Refinement on F^2	H-atom parameters constrained		
$R[F^2 > 2\sigma(F^2)] = 0.048$	$w = 1/[\sigma^2(F_o^2) + (0.0783P)^2]$		
$wR(F^2) = 0.138$	where $P = (F_o^2 + 2F_c^2)/3$		
S = 0.90	$(\Delta/\sigma)_{\rm max} < 0.001$		
2722 reflections	$\Delta \rho_{\rm max} = 0.15 \ {\rm e} \ {\rm \AA}^{-3}$		
199 parameters	$\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ \AA}^{-3}$		

Table 1

Selected geometric parameters (Å, °).

C1-C2	1.198 (3)	C5-C15	1.429 (3)
C1-C9	1.426 (3)	C6-C8	1.425 (3)
C2-C3	1.424 (3)	C6-C7	1.433 (3)
C3-C6	1.371 (3)	C7-N1	1.145 (3)
C3-C4	1.419 (3)	C8-N2	1.152 (3)
C4-C5	1.199 (3)		
C2-C1-C9	179.4 (2)	C4-C5-C15	177.2 (2)
C1-C2-C3	178.8 (3)	C3-C6-C8	121.53 (19)
C6-C3-C4	120.44 (19)	C3-C6-C7	121.1 (2)
C6-C3-C2	120.98 (19)	C8-C6-C7	117.4 (2)
C4-C3-C2	118.6 (2)	N1-C7-C6	179.1 (3)
C5-C4-C3	179.0 (2)	N2-C8-C6	179.5 (3)
C4-C3-C6-C8	-1.4 (3)	C2-C3-C6-C7	-1.8 (3)



Figure 2

Packing diagram of the title compound with view direction slightly rotated from the a axis. Hydrogen bonds are indicated by dashed lines. Radii are arbitrary.

Table 2

Hydrogen-bonding geometry (Å, °).

$D-\mathrm{H}\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdots A$
$C19-H19\cdots N1^{i}$	0.95	2.61	3.436 (3)	145
	1.1			

Symmetry code: (i) 1 - x, $y - \frac{1}{2}, \frac{1}{2} - z$.

H atoms were included using a riding model, starting from idealized positions.

Data collection: P3 (Nicolet, 1987); cell refinement: P3; data reduction: XDISK (Nicolet, 1987); program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP (Siemens, 1994); software used to prepare material for publication: SHELXL97.

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References

- Hopf, H., Kreutzer, M. & Jones, P. G. (1991). Angew. Chem. Int. Ed. Engl. 30, 1127–1128.
- Kreutzer, M. (1993). PhD thesis, Technical University of Braunschweig, Germany.
- Nicolet (1987). P3 and XDISK. Nicolet Instrument Corporation, Madison, Wisconsin, USA.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Siemens (1994). XP. Version 5.03. Siemens Analytical X-ray Instruments, Madison, Wisconsin, USA.
- Zhao, Y., Campbell, K. & Tykwinski, R. R. (2002). J. Org. Chem. 67, 336–344, and references therein.